

# Magneto-Optical Measurements of the Pigments in Fully Active Photosystem II Core Complexes from Plants<sup>†</sup>

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**ABSTRACT:** Preparation of a minimum PSII core complex from spinach is described, containing four Mn per reaction center (RC) and exhibiting high O<sub>2</sub> evolving activity [ $\sim 4000 \mu\text{mol of O}_2 (\text{mg of chl})^{-1} \text{h}^{-1}$ ]. The complex consists of the CP47 and CP43 chlorophyll binding proteins, the RC D1/D2 pair, the cytochrome *b*<sub>559</sub> subunits, and the Mn-stabilizing psbO (33 kDa) protein, all present in the same stoichiometric amounts found in the parent PSII membranes. Several small subunits are also present. The cyt *b*<sub>559</sub> content is 1.0 per RC in core complexes and PSII membranes. The total chlorophyll content is 32 chl *a* and <1 chl *b* per RC, the lowest yet reported for any active PSII preparation. The core complex exhibits the characteristic EPR signals seen in the S<sub>2</sub> state of higher plant PSII. A procedure for preparing low-temperature samples of very high optical quality is developed, allowing detailed optical studies in the S<sub>1</sub> and S<sub>2</sub> states of the system to be made. Optical absorption, CD, and MCD spectra reveal unprecedented detail, including a prominent, well-resolved feature at 683.5 nm (14 630 cm<sup>-1</sup>) with a weaker partner at 187 cm<sup>-1</sup> to higher energy. On the basis of band intensity, CD, and MCD arguments, these features are identified as the exciton split components of P680 in an intact, active reaction center special pair. Comparisons are made with solubilized D1/D2/cyt *b*<sub>559</sub> material and cyanobacterial PSII.

Photosystem II (PSII)<sup>1</sup> is the large multisubunit protein complex found in all oxygenic photosynthetic organisms. PSII catalyzes the oxidation of water to molecular oxygen, within the thylakoid membranes of higher plants and algae, as well as cyanobacteria (1–3).

Central to PSII is a membrane-incorporated core complex comprising the reaction center (RC) D1 and D2 proteins, the inner chlorophyll *a* (chl *a*) binding proteins CP43 and CP47, the cytochrome *b*<sub>559</sub> (cyt *b*<sub>559</sub>) heterodimer, and several small peptides (<10 kDa) of as yet unknown function (4). The D1/D2 heterodimer contains the photo-oxidizable special chlorophyll pair (P680) as well as the pigment components [two pheophytin *a* (pheo *a*), two accessory and two peripheral chl *a*] associated with primary light-driven charge separation across the thylakoid membrane. The D1 peptide appears to ligate the Mn cluster responsible for catalytic water oxidation and binds an extrinsic, water-soluble protein (33 kDa, psbO) at the luminal cytoplasmic face, which helps stabilize this Mn center (5).

Major differences between the PSII structures in higher plants and cyanobacteria involve the outer light-harvesting (LHC) protein assemblies and Mn-stabilizing peptides (6). Both higher plants and cyanobacteria have two extra (in addition to the 33 kDa protein) extrinsic, lower molecular weight Mn-stabilizing proteins (23 and 16 kDa proteins in plants, cyt *c*<sub>550</sub>, and PSII-U in cyanobacteria).

A number of procedures have been reported to isolate core complexes from cyanobacteria and plants. The detergent-solubilized species obtained exhibit varying degrees of functional (water splitting) intactness [e.g., see (7) and references cited therein, (8–18)]. Recently, PSII preparations from thermophilic cyanobacteria with high specific turnover activity [up to 5000  $\mu\text{mol of O}_2 (\text{mg of chl})^{-1} \text{h}^{-1}$ ]

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<sup>1</sup> Abbreviations: BBY, solubilized PSII membrane fragments; BIS-TRIS, [bis(2-hydroxyethyl)amino]tris(hydroxymethyl)methane; CD, circular dichroism; CP29/43/47, chlorophyll binding proteins of 29, 43, and 47 kDa molecular mass, respectively; chl, chlorophyll; cyt, cytochrome; D1/D2/b559, solubilized PSII containing the D1 and D2 proteins and cyt *b*<sub>559</sub>; DCMU, 3-(3,4-dichlorophenyl)-1,1-dimethylurea;  $\beta$ -d-DDM, *n*-dodecyl  $\beta$ -D-maltoside; EPR, electron paramagnetic resonance; FPLC, fast-performance liquid chromatography; HPLC, high-performance liquid chromatography; IR, infrared; LHC, light-harvesting complex; LiDS, lithium dodecyl sulfate; MCD, magnetic circular dichroism; MES, 2-(*N*-morpholino)ethanesulfonic acid; NIR, near-infrared; pheo, pheophytin; PPBQ, phenyl-*p*-benzoquinone; PSII, photosystem II; Q<sub>A</sub>, primary quinone acceptor; Q<sub>B</sub>, secondary quinone acceptor; RC, reaction center; TRIS, tris(hydroxymethyl)aminomethane; VB, visual basic.

have been successfully crystallized (17–20).

Comparably purified PSII core complex material from higher plants has often shown much lower specific activity [ $<1000 \mu\text{mol of O}_2 (\text{mg of chl})^{-1} \text{ h}^{-1}$ ] (9, 11, 12), unless some of the LHC components are retained (12, 21). Until recently, even the chlorophyll stoichiometry of the PSII core complex was uncertain. Earlier estimates suggested up to  $\sim 60$  chl *a* per reaction center (1). It is now clear from both pigment analysis (10, 12, 17), and particularly from medium-resolution 2D and 3D crystallography on higher plant (22, 23) and cyanobacterial systems (20), that the total chl *a* content of the core complex (i.e., of CP47+CP43+D1/D2 reaction center) is 32–35 chl *a*, with 2 pheophytins *a* per reaction center.

The protein–pigment organization of the D1/D2 region of PSII is now known to closely resemble the purple bacterial reaction center structure (20, 22, 24, 25). Nevertheless, a detailed interpretation of the electronic structure and interaction of the pigment components of the PSII reaction center is still controversial [e.g., (26–29)]. This is mainly due to a lack of distinct features within the relatively compact spectral region identified with the PSII reaction center, as compared to spectra of the bacterial system, where interaction splittings are much greater.

Most optical studies addressing the dynamics of energy transfer and pigment coupling in PSII have employed highly solubilized D1/D2/cyt *b*<sub>559</sub> particles (D1/D2/*b*<sub>559</sub>) (30). Although functional in charge separation, the reaction center environment in these preparations is heterogeneous (31) and substantially altered from its native, in vivo state, having lost all peripheral membrane and cytoplasmic protein components as well as the Mn cofactors.

Recently, we have embarked on an extensive program of variable-temperature absorption, emission, circular dichroism (CD), and magnetic circular dichroism (MCD) spectroscopy of enzymatically functional PSII core complexes from higher plants and cyanobacteria. This required the development of preparative protocols for functionally intact PSII core complexes from higher plants having a minimum LHC pigment contamination. It also required the evolution of techniques by which samples of high optical clarity and isotropy could be prepared at low temperatures.

Here we describe this preparative procedure, which is the first for PSII core complexes from higher plants to exhibit very high catalytic activity levels, comparable to those attainable from thermophilic bacterial sources. Low-temperature ( $<10$  K) optical absorbance studies show that the chl *a* Q<sub>y</sub> region displays very well resolved pigment features, significantly narrower than those observed in D1/D2/*b*<sub>559</sub> preparations.

## MATERIALS AND METHODS

**PSII Sample Preparation.** In basic outline, the present procedure for PSII core complex preparation from spinach follows established methods. Its unique success appears to derive from careful attention to a number of details in the total process, from PSII membrane preparation through to column separation of the solubilized core complex material. The key features are all given here.

All procedures of PSII membrane and core complex preparation were undertaken under dim green light and at 4

°C unless otherwise indicated. PSII membrane samples were prepared according to Smith et al. (32). Protease inhibitor was not used. Bovine serum albumin (BSA fraction IV, Sigma) was added as a sacrificial protein to all buffers prior to the Triton X-100 detergent solubilization step. The solubilized membrane samples were washed once in 50 mM MES, pH 6.0, 50 mM NaCl, and 5 mM MgSO<sub>4</sub> to remove Triton detergent. The solubilized membrane samples were stored at 12–16 mg of chl/mL in a buffer containing 0.4 M sucrose, 10 mM NaCl, 15 mM MgCl<sub>2</sub>, and 20 mM MES, pH 6.0 (NaOH), at  $-85^\circ\text{C}$ .

Preparation of PSII core complexes followed, in principle, the procedure of van Leeuwen et al. (9). All buffers used in the PSII core complex preparation were prefiltered using a 0.45  $\mu\text{m}$  filter. This removed particulates, which may affect the FPLC process. The PSII membrane samples ( $\sim 30$ –40 mg of total chlorophyll) were diluted to 2.0 mg/mL in a system buffer consisting of 0.4 M sucrose, 20 mM MgCl<sub>2</sub>, 10 mM MgSO<sub>4</sub>, 5 mM CaCl<sub>2</sub>, 0.3 mg/mL  $\beta$ -d-DDM, and 20 mM BIS-TRIS, pH 6.5 (HCl). The diluted sample was solubilized by stirring for 10 min at 25 °C in darkness with  $\beta$ -d-DDM (as a 10% detergent solution in system buffer) to a final concentration of  $\sim 71.5$  mg of  $\beta$ -d-DDM/mg of chl (1.25%  $\beta$ -d-DDM w/v in the solubilization). The solubilized membranes were centrifuged at 17000g at 15 °C for 15 min. The supernatant was then placed on ice in darkness to halt solubilization. The PSII core isolation procedure followed the perfusion chromatography protocol according to Roobal-Bóza et al. (11) using a Poros HQ 20 column medium, self-packed to a pressure of  $<2$  MPa. Flow rates were maintained at about 2 column volumes per minute while maintaining a column backpressure below 2 MPa. The PSII core complexes were eluted following the initial LHC peak with system buffer plus 80 mM MgSO<sub>4</sub>, using a 2 min linear gradient from the system buffer to the elution buffer. The eluted fractions containing the PSII core complexes were pooled and concentrated using Centriprep centrifugal concentrators (Amicon, 10 kDa MW cutoff) and stored (1–3 mg of chl/mL in the elution buffer) at  $-85^\circ\text{C}$  until used. If more than one column run was undertaken, each FPLC procedure was run after column regeneration consisting of 15 column volumes of 1 M NaCl, followed by 15 column volumes of ultrapure H<sub>2</sub>O, followed by the preequilibration step of 15 column volumes of system buffer. The yield of chl in the eluted PSII core complex samples was 3–5% of the total chl in the starting PSII membranes prior to solubilization.

All chl content and chl *a/b* ratios were determined using buffered 80% acetone as a solvent, following the protocol of (33).

**Oxygen Evolution.** Oxygen evolution measurements employed a Hansatech electrode system with PPBQ and K<sub>2</sub>Fe(CN)<sub>6</sub> (both 200  $\mu\text{M}$ ) as electron acceptors. Acceptor side intactness was examined by addition of DCMU (50  $\mu\text{M}$ ) to the oxygen electrode measurement chamber.

**Pigment and Metal Analysis.** PSII samples were diluted to 40  $\mu\text{M}$  chl concentration in buffered 80% acetone solution (HPLC grade, pH 7.8) and centrifuged at 10000g to remove particulates and detergent. HPLC pigment analysis was undertaken according to (34).

Mn determination was performed by acidifying the PSII material in 1 M HCl and quantitating the resulting Mn '6 line' electron paramagnetic resonance (EPR) signal at 10 K

against EPR standards. This also allowed an estimate of the unbound Mn content (i.e., EPR-visible species) in the intact PSII material (see Figure 5).

**Polypeptide Analysis.** Polyacrylamide gel electrophoresis employed a 10–17% polyacrylamide gradient gel system (30) with 6 M urea in the resolving gel and peptide solubilization buffer and Coomassie Blue staining. Western blot analysis and antibody staining were carried out by T. Hundall and C. Spetea (Linköping University, Sweden) using procedures established in their laboratory (11).

**EPR Spectroscopy.** X-band EPR experiments were carried out on a Bruker ESP300E spectrometer using a microwave frequency of 9.42 GHz (35). Buffer and sample conditions for the PSII core samples used are described in the figure legends. Samples were examined with dilution in both system (10 mM MgSO<sub>4</sub>) and elution (90 mM MgSO<sub>4</sub>) buffers. No differences in the turnover capabilities of the EPR samples were observed based on the presence of up to 90 mM MgSO<sub>4</sub>. The presence of 30 mM Cl<sup>−</sup> in the elution buffer appears to protect the PSII core samples from SO<sub>4</sub><sup>2−</sup> inhibition with the 17 and 23 kDa extrinsics removed during the core isolation procedure.

Samples were illuminated for 12–20 s in a N<sub>2</sub> flow cryostat with a 250 W lamp at 255 K, using blue or green filtered light and a 10 cm water path IR filter, to induce S<sub>1</sub>-to-S<sub>2</sub> turnover (generating the multiline form of the S<sub>2</sub> state). The S<sub>2</sub> samples were subsequently illuminated at 145 K using a 715 nm cutoff filter in the far-red/near-infrared (NIR) for 15 min to generate the  $g \approx 4.1$  form of the S<sub>2</sub> state. The terminal acceptor site of the PSII cores prepared by the van Leeuwen (9) protocol is known to be unoccupied, with Q<sub>B</sub> missing. The illumination temperature allows full turnover of the PSII cores from the S<sub>1</sub> to the S<sub>2</sub> state while minimizing formation of carotenoid or chlorophyll radicals. These were observed to be generated using illumination temperatures below ~235 K (data not presented). By quantitation of the S<sub>2</sub> multiline EPR signal, sample turnover was observed to be 0.8–1.0 multiline spin formed per RC using the >255 K illumination protocol. Spin quantitation was done by double integration of S<sub>2</sub> – S<sub>1</sub> difference spectra, via comparison against the signal from Y<sub>D</sub><sup>ox</sup> and EPR standards (CuSO<sub>4</sub>).

**Optical Spectroscopy.** Thawed samples were diluted to ~0.3 mg of chl/mL in the elution buffer and dark-adapted at 20 °C for 2 min. Then 1:1 ethylene glycol/glycerol was added to give a final polyalcohol concentration of 40% (v/v), and the sample was syringed into a quartz windowed cell (~100 μL) of path length 1 mm. This was mounted onto a sample rod and lowered into an Oxford Instruments Spectromag 4 (SM4) cryostat. Glasses of good optical quality were obtained by cooling the solution (over approximately 40 s) from room temperature to 4 K. The absorption and CD, or the absorption and MCD, were collected *simultaneously* on an apparatus designed and constructed in our laboratory (36). The MCD measurements were collected upon application of a 5 T magnetic field to the sample.

## RESULTS

**PSII Core Complex Analysis.** Table 1 summarizes the analytical data on our PSII core complex material. The observed range in catalytic activity generally reflects variation in the activity of the starting PSII membrane material.

Table 1: Activity/Component Analyses<sup>a</sup> for Core Complexes and PSII Membrane Particles

	core complex	PSII membranes
activity [ $\mu\text{mol of O}_2$ (mg of chl) <sup>−1</sup> h <sup>−1</sup> ] + 50 $\mu\text{M}$ DCMU	3700–4300 < 5	600–800 ~ 0
component		
pheophytin <i>a</i>	2.00	2.00
chlorophyll <i>a</i>	32 ± 1	134 ± 2
chlorophyll <i>b</i>	0.8 ± 0.2	62 ± 1
$\beta$ -carotene	7 ± 1	9 ± 1 <sup>d</sup>
Mn	4.5 ± 0.5 <sup>b</sup>	~5
cyt <i>b</i> <sub>559</sub>	1.0 ± 0.05 <sup>c</sup>	1.0 ± 0.2 <sup>e</sup>

<sup>a</sup> All component stoichiometries are given per 2.00 pheo *a*, assumed to represent one PSII reaction center. <sup>b</sup> Bound content, excluding superficially associated Mn<sup>2+</sup> (see Figure 5, Materials and Methods).

<sup>c</sup> Determined by quantitative scaling of the MCD spectra (see text).

<sup>d</sup>  $\beta$ -Carotene only. Excludes xanthophylls and lutenes. <sup>e</sup> Determined by an analysis of the MCD spectra of PSII membranes and the relative intensities of the cyt *b*<sub>559</sub> subunit to core complex bands in gels (see text).

Rates of order 4000  $\mu\text{mol of O}_2$  (mg of chl)<sup>−1</sup> h<sup>−1</sup> are significantly higher than any previously reported for higher plant derived PSII material. This activity was almost completely inhibited by 50  $\mu\text{M}$  DCMU. We find a total of 32 chl *a* and 7 carotenoids ( $\beta$ -carotene only) per reaction center, assuming the standard 2.00 pheo *a* stoichiometry for the latter. In addition, we find 4.5 ± 0.5 Mn per 32 chl *a*, which is substantially higher than the generally reported Mn/chl ratio in cyanobacterial core complex preparations (7, 8, 10, 14).

A value of 32 chl *a* per reaction center is very close to the number observed (or inferred) from crystallographic studies (20, 22, 23) on higher plant or cyanobacterial core complexes (i.e., CP47, 14 chl; CP43, 12–14 chl; RC, 6 chl; 32–34 chl total). The carotenoid content (~7 per reaction center) is virtually the same as that reported by Hankamer et al. (12) for dimeric PSII core complexes. We presume that our material is in this form, although this has not been directly determined.

We find a single cyt *b*<sub>559</sub> per 32 chl *a*, by low-temperature MCD (see below). The oxidized and reduced forms are easily distinguished. The cytochrome is >95% oxidized in the buffer/cryoprotectant medium used to glass the core samples used for optical study. A single cyt *b*<sub>559</sub> per reaction center is consistent with earlier determinations on cyanobacteria (8, 10) and the recent PSII crystal structure for *S. elongatus* (20).

Figure 1 shows a comparison of gel patterns for detergent-solubilized core complex and the parent PSII particles. Notable is the near-absence of any LHC chl pigment proteins in the core complex, but the essentially quantitative retention (relative to the CP43 and CP47 bands) of the 33 kDa Mn-stabilizing protein. The 23 and 16 kDa extrinsic proteins are absent, as expected. Comparison with the gel patterns from earlier reported higher plant derived PSII core complex preparations [e.g., (12)] suggests that the high activity of the present material derives from its retention of the 33 kDa protein in functionally attached form.

Western blotting and antibody staining indicate that the faintly discernible feature below 30 kDa apparent molecular mass in the core complex gel pattern contains CP29. This contamination is <10 mol % (per complex) based on the chl *b* content (Table 1) and absorption spectra (Figure 2



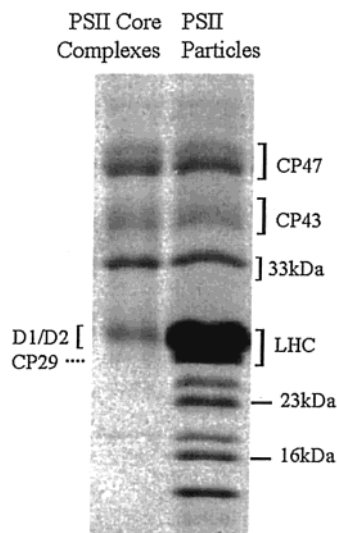


FIGURE 1: Gel comparison of peptide bands (above  $\sim 12$  kDa) from PSII core complexes and the PSII membrane material from which the complexes were made. The gels were loaded with approximately equal reaction center protein content. The CP43 band is broad, as is frequently seen in higher plant PSII [e.g., (12, 21)], possibly due to variable phosphorylation levels. Essentially stoichiometric retention of the 33 kDa Mn stabilizing protein in the core complexes is evident, with total loss of the 16 and 23 kDa extrinsic proteins. The band labeled CP47 shows two components of slightly differing apparent molecular mass. Both are confirmed to contain CP47 by peptide fragment sequencing. This behavior of CP47 has been seen previously (69). A comparison of Figure 1 above and Figure 5 in (69) suggests that the relative distribution of CP47 between the two forms is variable.

below). Low molecular weight gels and blotting (not shown) also indicate the presence of several small PSII subunits, including psbW and the  $\alpha$  subunit of cyt  $b_{559}$ .

**Absorbance and CD.** Figure 2a shows the visible–NIR absorption spectrum of the PSII core complex in the dark-adapted ( $S_1$ ) state at 1.5 K. The pheophytin  $Q_x$  region is shown in detail as an inset. The inset shows a combination of active (pheo 1) and inactive branch (pheo 2) chromophore bands, which are distinguished by differential electrochromic shifts following light-induced turnover to the  $Q_A^-S_2$  state (37).

Figure 2b shows the chlorophyll  $Q_y$  region in detail, together with the corresponding CD spectrum. The prominent feature at  $14\,630\text{ cm}^{-1}$  (683.5 nm) has a line width of only  $45\text{ cm}^{-1}$  and a distinct negative CD band. The same feature is present in the absorption and CD of intact PSII particles, with approximately the same intensity (per reaction center) and line width (Figure 2c).

The absorption envelope in Figure 2b consists of two partially overlapping regions: a broad approximately Gaussian band centered around  $14\,900\text{ cm}^{-1}$  ( $\sim 670$  nm) which is responsible for most of the total intensity (70–80%) and a spectrally compact but structured region toward the red which includes the  $14\,630\text{ cm}^{-1}$  band. The structure observed in this region is more resolved than that seen in D1/D2/ $b_{559}$  particles [e.g., (28–31, 38, 39)] or isolated CP43, CP47 complexes [e.g., (40–42)].

Consistent with the gel pattern results in Figure 1, a comparison of the  $Q_y$  absorption and CD envelopes in Figure 2b,c shows little, if any, detectable LHC spectral contribution in the core complexes. This is particularly evident in the CD

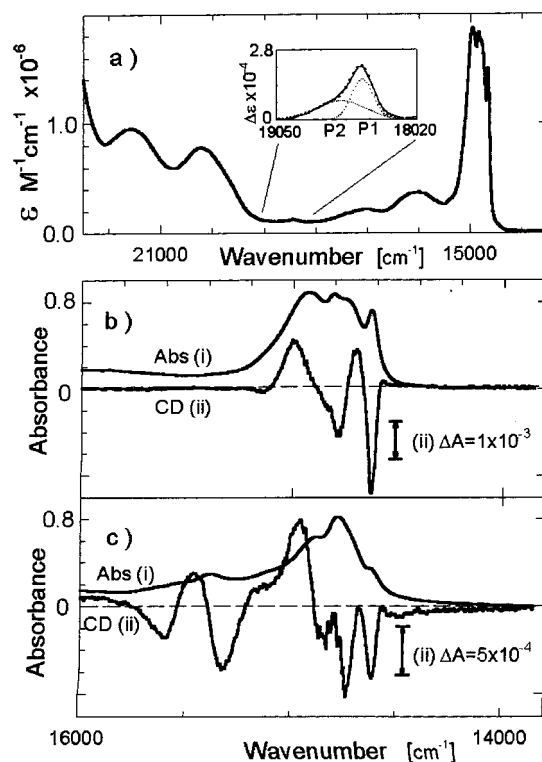


FIGURE 2: (a) Visible/NIR region absorption spectrum of PSII core complexes. The inset shows the pheophytin  $Q_x$  region, with active (P1) and inactive (P2) branch chromophore bands indicated. The extinction coefficient (per reaction center) scale assumes a dipole strength,  $D$ , for the chl  $a$   $Q_y(0,0)$  band of  $19.1\text{ D}$  (44) and pigment compositions from Table 1. (b, c) Absorbance and CD spectra of the chlorophyll  $Q_y$  region of core complexes and PSII particles, respectively. All spectra are recorded at 1.9 K in a glassed matrix as described under Materials and Methods.

spectra, where LHC chl  $b$  contributions in the  $15\,000$ – $16\,000\text{ cm}^{-1}$  region are below the noise level in the CD spectra of the core complexes.

**MCD.** Figure 3a shows the visible/NIR absorption and MCD spectra of the PSII core complexes at 1.5 K. The MCD spectrum is the first reported for functional oxygen-evolving PSII material. It is dominated by the chl  $a$  Faraday  $B$ -term (diamagnetic and temperature-independent) (43). The sign is positive for the band origin  $Q_y(0,0)$  transition and negative for the  $Q_y(0,1)$  (44). For the  $Q_y(0,0)$  band of an isolated chl  $a$  chromophore, the characteristic property  $B_0/D_0$  (MCD moment/transition moment) is proportional to  $\Delta\epsilon_M/\epsilon$  (molar ellipticity/extinction coefficient), and the latter has been determined to be  $5.53 \times 10^{-4}\text{ T}^{-1}$  (44).

Figure 3b shows detailed absorption and MCD spectra in the chl  $Q_y$  region. The observed spectra have been scaled (according to the above  $\Delta\epsilon_M/\epsilon$  ratio for a 5 T field) to display a marked variation in the magnitude of the MCD [Figure 3b(ii)], compared to absorption [Figure 3b(i)], across the band.

For isolated chl  $a$ , the  $Q_y$  band shapes in absorption and MCD are the same to a good approximation (44). In a distribution of noninteracting chl  $a$  chromophores, the  $\Delta A(\text{MCD})$  will be proportional to the absorbance,  $A$ , at any point. In the high-energy region of the core complex  $Q_y$  envelope, the chl  $a$  do indeed behave, with respect to their MCD, as a collection of monomeric chromophores. However, the (lower energy) reaction center region displaying resolved

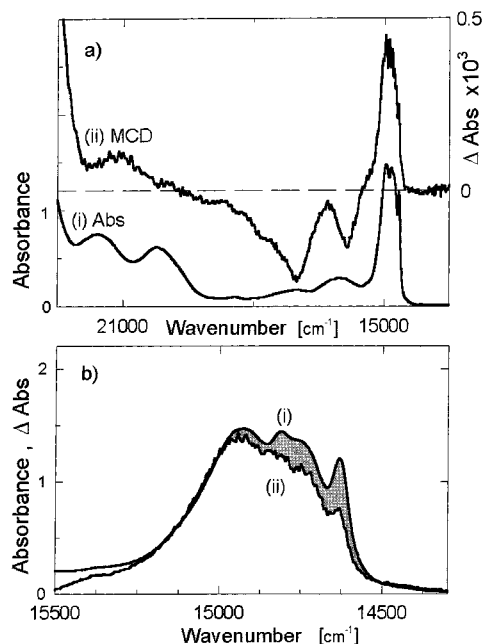


FIGURE 3: (a) Comparison of absorption and MCD spectra for PSII core complexes over the same observation range as in Figure 2a. Sharp derivative features near  $18\,000\text{ cm}^{-1}$  are undetectable, indicating an essential absence of the reduced form of cyt  $b_{559}$  (see text). (b) Comparison of absorbance and MCD spectra in the chlorophyll  $Q_y$  region. The MCD intensity has been scaled as described in the text. The filled area corresponds to the apparent MCD deficit (see text). The temperature is 1.5 K, field 5.0 T. Sample conditions as in Figure 2.

structural features exhibits a significantly smaller  $\Delta A/A$  ratio compared to isolated chl  $a$ . This ‘MCD deficit’ (filled region in Figure 3b) is most pronounced at the prominent  $14\,630\text{ cm}^{-1}$  ( $683.5\text{ nm}$ ) feature, where the  $\Delta\epsilon_M/\epsilon$  has fallen to approximately half that for monomeric chl  $a$ .

Some of the observed MCD deficit may originate from pheophytin. We note that the peak  $\Delta\epsilon/\epsilon$  for (isolated) pheo  $a$  is  $\sim 40\%$  that of (isolated) chl  $a$ . However, the total deficit observed in Figure 3b corresponds to a total of 3–4 chl  $a$  molecules. As the absorption strength (in integrated intensity) of pheo  $a$  in the  $Q_y$  region is 74% that of chl  $a$ , it follows that 2 pheo  $a$  can account for only  $\sim 1.4$  chl  $a$  deficit equiv. The remaining deficit we attribute to a reduction of chl  $a$  MCD associated, phenomenologically, with interchromophore coupling. This is discussed further below.

Other features readily observable in the low-temperature MCD spectra are those associated with the heme  $\alpha$ -band region ( $17\,000\text{--}18\,000\text{ cm}^{-1}$ ,  $550\text{--}560\text{ nm}$ ) of low-spin cyt  $b_{559}$ , in both the  $\text{Fe}^{\text{II}}$  and  $\text{Fe}^{\text{III}}$  forms (Figure 3a).  $\text{Fe}^{\text{II}}$  cytochrome has a strong, sharp Faraday A-term feature in the  $\alpha$ -band region, with a characteristic derivative shape (45). In the solubilized PSII core complexes studied here, such derivative-shaped signals from the reduced form are not seen (Figure 3a), establishing that  $>95\%$  of the cytochrome is oxidized. (This is supported by EPR; see Figure 5 and below.) A  $\text{Fe}^{\text{III}}$  MCD band is temperature-dependent, displaying  $1/T$  temperature dependence (Faraday C-term behavior). This is distinctly different from that associated with diamagnetic species absorbing in the region (chl, pheo, and  $\text{Fe}^{\text{II}}$  cytochrome).

Figure 4 shows the temperature-dependent MCD spectra of the PSII core complex over the same wavelength range

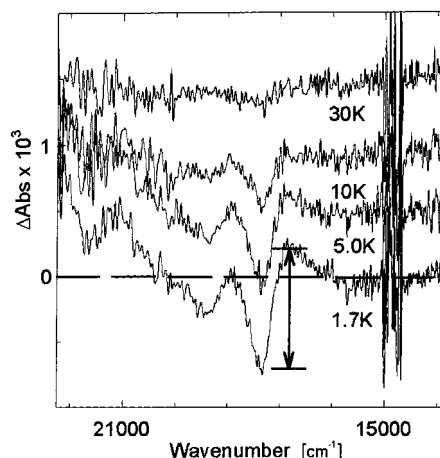


FIGURE 4: Temperature-dependent component of MCD spectra of PSII complexes over the frequency range of Figure 2a. The MCD presented is that at the observation temperature minus that at 60 K. Increased noise in the chl  $Q_y$  region is inherent and arises from the low light levels in regions of high absorbance. Field 5 T and sample conditions as in Figure 2. The bar shows points between which  $\delta\Delta\epsilon_M$  was measured (see text). The extinction coefficient scale is established as described in Figure 2, from simultaneously acquired MCD and absorption data.

as in Figures 2a and 3a. The spectra closely resemble those of  $\text{Fe}^{\text{III}}$  heme proteins in all relevant features (45). MCD spectral features around  $17\,000\text{--}18\,000\text{ cm}^{-1}$  show relatively little variation in  $\text{Fe}^{\text{III}}$  heme proteins, in both shape and absolute intensity (45–48). The  $\delta\Delta\epsilon_M$  value between the  $17\,400\text{ cm}^{-1}$  trough and the  $16\,800\text{ cm}^{-1}$  peak (at 4 K and 5 T, see Figure 4) varies only weakly with heme ligand type [ $\delta\Delta\epsilon_M = 1.1 \times 10^3\text{ M}^{-1}\text{ cm}^{-1}$  (mono his),  $1.3 \times 10^3\text{ M}^{-1}\text{ cm}^{-1}$  (his-his),  $(1.3\text{--}1.5) \times 10^3\text{ M}^{-1}\text{ cm}^{-1}$  (his-met)]. Although cyt  $b_{559}$  has never itself been measured, the spectral shape in Figure 4 is indistinguishable from that of the bi-his-ligated cyt  $c3$  center (46), for which  $\delta\Delta\epsilon_M = 1.30 \times 10^3\text{ M}^{-1}\text{ cm}^{-1}$  (4.2 K and 5 T). Using this value, the cyt  $b_{559}$ /reaction center stoichiometry given in Table 1 is obtained.

In analogous MCD spectra of PSII membrane particles (not shown), both reduced and oxidized cytochrome signals are seen. Again using literature MCD  $\Delta\epsilon_M$  values (45), the total cytochrome content sums to one per reaction center, but the balance between the oxidized and reduced fractions is variable between membrane samples.

**EPR.** Functionally intact PSII exhibits two well-known low-temperature EPR signals arising from the Mn catalytic site: the multiline signal centered at  $g \approx 2$  and a signal at  $g \approx 4.1$ . Both are formally associated with the  $S_2$  state (5). The  $S_2$  state is obtained by single electron turnover from the dark-stable  $S_1$  state. Although EPR signals from all four ( $S_0$ ,  $S_1$ ,  $S_2$ ,  $S_3$ ) quasi-stable intermediates of the Mn cluster have now been identified (49), the  $S_2$  multiline and  $g \approx 4.1$  signals are the best characterized and are easily generated by well-established protocols of low-temperature continuous illumination.

Figure 5 shows the low-temperature EPR spectra of the core complex in the  $S_1$  (dark state) and  $S_2$  states. The  $S_1$  state spectrum clearly shows signals at  $g \approx 2.26$ ,  $2.97$  arising from low-spin  $\text{Fe}^{\text{III}}$  cyt  $b_{559}$ , as well as background rhombic  $\text{Fe}^{\text{III}}$  ( $g \approx 4.3$ ) normally present in photosystem preparations (50). Green or blue illumination (see Materials and Methods)

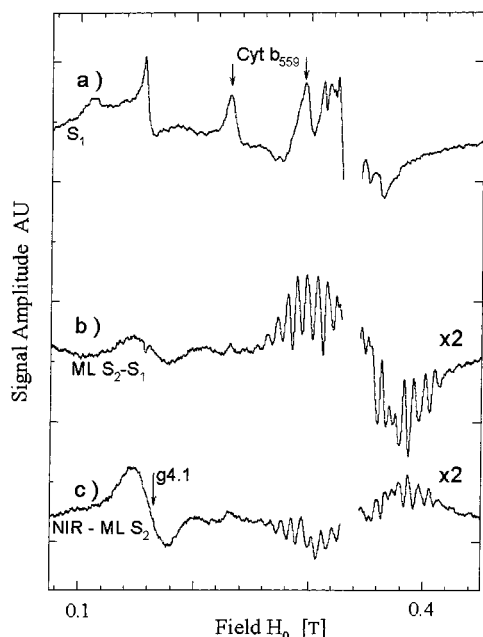


FIGURE 5: EPR spectra of PSII core complex material in the same cryoprotectant/buffer used for optical studies (see Materials and Methods, 1 mg of chl/mL final sample concentration). (a)  $S_1$  represents the dark state sample with downfield peaks of oxidized, low-spin cyt  $b_{559}$  as indicated (arrows). Peaks with 9 mT spacing in the  $g = 2$  region are from unbound  $Mn^{2+}$  and correspond to  $\sim 0.3 Mn^{2+}$  per reaction center. The radical region close to  $g = 2$  [mainly oxidized tyrosine,  $Y_D^{ox}$  (5)] has been omitted. (b)  $ML S_2 - S_1$  is the difference spectrum for the multiline form of the  $S_2$  state minus the  $S_1$  state (see Materials and Methods). (c)  $NIR - ML S_2$  is the difference spectrum corresponding to partial conversion ( $\sim 50\%$ ) of the multiline to the  $g \approx 4.1$  form of the  $S_2$  state by low-temperature NIR illumination (see Materials and Methods). The multiline region has negative phase, showing loss of signal. Vertical scaling of (b) and (c) is twice that of (a). EPR conditions: modulation amplitude, 20 G; modulation frequency, 100 kHz; microwave frequency, 9.42 GHz; microwave power, 5 mW; temperature, 9 K.

generates a multiline signal that is very similar to that seen in PSII membranes (35). Integration (see Materials and Methods) of this signal indicates that approximately 1 spin = 1/2 center has been formed per 32 chl (i.e., per RC). NIR illumination of PSII poised in the  $S_2$  spin = 1/2 state giving rise to the multiline signal converts the Mn center into the higher spin state giving rise to the  $g \approx 4.1$  signal (51, 52). Figure 5 shows that this procedure works efficiently in the PSII core complex material, with 40–70% NIR-induced interconversion between the multiline and  $g \approx 4.1$  states being achievable at 140 K illumination. No clear evidence is seen for the signals at higher  $g$  ( $g \approx 8-10$ ) reported for NIR illumination below 150 K (52).

Although the EPR characterization given here is for PSII material glassed in the polyalcohol cryoprotectant medium (40%) necessary for the optical studies, essentially the same behavior is seen for the cytochrome and Mn signals when the complexes are frozen in 0.4 M sucrose buffer (not shown). The metal centers (particularly Mn) appear to become perturbed only at polyalcohol levels above  $\sim 50\%$  v/v.

## DISCUSSION

The recent medium-resolution structural determination of the PSII membrane complex from cyanobacteria (20) is

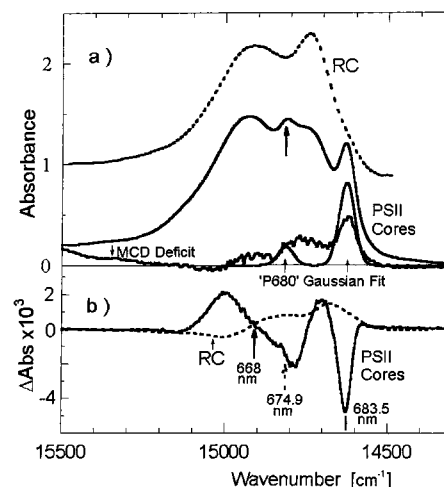


FIGURE 6: (a) Comparison of chl  $Q_y$  region for PSII core complexes (this work) and a typical D1/D2/ $b_{559}$  preparation (38). Inequivalent vertical scales. With the core complex spectrum are shown the scaled MCD deficit spectrum from Figure 3b (absorbance minus monomer-scaled MCD intensity, see text) and two equal-width (45  $cm^{-1}$ ) Gaussian bands centered at 14 630  $cm^{-1}$  (683.5 nm) and 14 817  $cm^{-1}$  (674.9 nm), which are fits to the partially resolved features in the core complex absorption envelope assigned to the  $Q_y^{\pm}$  components of the exciton split P680 pair (see text). The total area of the 14 630  $cm^{-1}$  Gaussian corresponds to 2.2 chl  $a$ , assuming the extinction scale from Figure 2a. The midpoint of the  $Q_y^{\pm}$  components, 14 724  $cm^{-1}$  (679.2 nm, i.e., the mean apparent monomer transition energy), is very close to the P680 band center seen in D1/D2/ $b_{559}$  by absorption, MCD, and triplet-minus-singlet difference spectra (38, 44, 60). The exciton splitting,  $\sim 190 cm^{-1}$ , is close to the estimated width ( $\sim 200 cm^{-1}$ ) of the single, inhomogeneously broadened P680 band seen at 4 K in D1/D2/ $b_{559}$  by hole-burning (38). (b) Comparison of CD spectra for the samples from (a). The relative scaling of the CD to absorbance is the same for the core complexes and D1/D2/ $b_{559}$ . Arrows indicate the approximate centers for the broad  $\sim 670$  nm Gaussian band assigned to mainly CP43, CP47 (see text), and the small positive CD feature corresponding to the 674.9 nm band inferred in (a) (see text).

expected to significantly advance our understanding of this photosystem. However, many issues with respect to function remain to be clarified. Also, PSII derived from eukaryotes or higher plants may differ in a number of significant ways from cyanobacterial PSII. In addition to the gross differences associated with the light-harvesting and Mn-stabilizing extrinsic proteins, it is likely that there are significant differences, at the detailed structural level, between the reaction center pigment organization in the two systems (see below). Magnetic interactions within the catalytic site Mn clusters may also differ.

In terms of spectral detail, the present material provides the most detailed 'view' yet obtained for intact, oxygen-evolving PSII from any source. Figure 6 shows a comparison of the chl  $a$   $Q_y$  regions for the PSII core complexes and typical spectra for D1/D2/ $b_{559}$  material from the literature (38). In Figure 6a, the scaled MCD deficit (total absorbance minus MCD intensity, scaled for monomeric chl  $a$  as previously outlined) from Figure 3b is shown, as well as a least-squares fit to a Gaussian of the distinct 14 630  $cm^{-1}$  feature in the core complex. We can discern several components in the  $Q_y$  region of the spectra:

(1) The MCD deficit is significant only below  $\sim 14 900 cm^{-1}$  ( $> 670$  nm), indicating substantial coupling between the pigments in this region. This is the region also exhibiting the most structured absorbance and CD behavior. This region



is most likely dominated by the more strongly coupled reaction center pigments.

(2) The CD spectrum, which is featureless in the range 16 000–21 000  $\text{cm}^{-1}$ , is intense and close to conservative in the  $Q_y$  region. This is expected if exciton coupling is the principal mode of pigment interaction in the core complexes and that this process dominates the CD. The CD of monomeric chl *a* is weak, negative, and closely follows the absorption shape with  $\Delta\epsilon_{\text{max}}/\epsilon_{\text{max}} = -0.16 \times 10^{-3}$  (53). In the blue region (above 14 700  $\text{cm}^{-1}$ ) of the core complex spectrum, which consists mainly of a broad, quasi-Gaussian absorption feature [center  $\sim 14\,930\text{ cm}^{-1}$  (670 nm), width  $\sim 250\text{ cm}^{-1}$ ], the CD has a maximum CD/peak absorption,  $|\Delta\epsilon_{\text{max}}/\epsilon_{\text{max}}|$ , of  $\sim 2 \times 10^{-3}$ . The near to simple derivative shape is similar to that observed in light-harvesting proteins (54). The CD zero crossing is only slightly red-shifted ( $\sim 10$ – $20\text{ cm}^{-1}$ ) from the maximum. This may indicate that the spectroscopic environment of the responsible chromophores is similar (54). The broad Gaussian accounts for  $\sim 70\%$  of the total  $Q_y$  intensity. We assign it to be, principally, the CP47 and CP43 inner light-harvesting proteins in the assembled complex.

There are literature reports of CD from oxygen-evolving PSII material [cores and BBY (55), and chloroplast fragments (56)]. These spectra have some similarity to those of the present core preparation, with a prominent negative feature around 14 600  $\text{cm}^{-1}$  and a positive feature near 15 000  $\text{cm}^{-1}$ , but they also have marked features below 16 000  $\text{cm}^{-1}$  (55), where our material shows no CD. The reported CD spectra from isolated CP43 (39) and CP47 (40), when added, resemble the present core complex  $Q_y$  CD in the blue region (above 14 700  $\text{cm}^{-1}$ ) but differ significantly in the red region, lacking the intense 14 630  $\text{cm}^{-1}$  feature.

(3) There is a significant absorbance tail to the red of the 14 630  $\text{cm}^{-1}$  band. This corresponds to  $\sim 1$  chl centered at  $\sim 14\,600\text{ cm}^{-1}$  (685 nm) with peak width  $< 100\text{ cm}^{-1}$ . This band does not exhibit an MCD deficit and so appears not to be part of any strongly coupled system. Bands corresponding to 1–2 chl in total intensity have been observed in this region for isolated CP47 (41).

*The Spectroscopic Signature of P680.* Within the spectral region assigned to the reaction center pigments (those displaying an MCD deficit, Figure 3b), the most prominent feature is the peak at 14 630  $\text{cm}^{-1}$  (683.5 nm). The band area of the Gaussian fit to this absorption peak corresponds to 2.2 chl *a* in total  $Q_y(0,0)$  dipole strength, and has a width of only 45  $\text{cm}^{-1}$ . Two recent results from intact PSII systems include both P680 triplet-minus-singlet and P680<sup>+</sup>Pheo<sup>−</sup> – P680Pheo difference spectra (57–59). These show a bleaching band at 684–685 nm, centered at the same wavelength as the strong 683.5 nm feature seen here. The line width obtained from the P680<sup>+</sup>Pheo<sup>−</sup> – P680Pheo bleaching in (58) ( $\sim 60\text{ cm}^{-1}$ ) is also comparable to that which we observe, although the line width obtained from the triplet difference spectra (57, 58) is somewhat greater.

We can then associate the clearly resolved absorption feature in Figure 6 with P680 in higher plants. The total oscillator strength of  $\sim 2.2$  chl *a* equiv suggests that at least 2 chl *a* are involved.

As mentioned previously, the MCD deficit at the 14 630  $\text{cm}^{-1}$  (683.5 nm) feature is  $\sim 50\%$ . We note that the special pair reaction center dimers of PSI (P700), purple bacteria

(P850), and PSII in the form of D1/D2/*b*<sub>559</sub> complexes (possibly perturbed P680) each exhibited  $B_0/D_0$  values close to half the respective monomer values. Phenomenologically, this deficit is a signature of interchromophore coupling. Nozawa and colleagues have shown (44, 60) that the  $B_0/D_0$  for bacteriochlorophyll in solution is dependent on the degree of aggregation. For dimers, the  $B_0/D_0$  value was 50% of the monomer value, becoming progressively less in higher aggregate (trimer, tetramer) forms. Assuming that the MCD deficit of the P680 pigments arises from interchromophore coupling, the 50% MCD value observed implies predominantly *pairwise* coupling within the P680 moiety.

We are presently engaged in a detailed theoretical study of the spectral consequences of a pigment interaction scheme dominated by an exciton-coupled special pair. One interesting preliminary finding (J. L. Hughes et al., in preparation) is that couplings between the central chlorophylls and each of the accessory chl pigments can be as large as  $\sim 30\%$  of the special pair coupling (i.e.,  $\sim 30\text{ cm}^{-1}$ ), *without* significantly perturbing the dimer-like spectral appearance of the central pair. Given the characteristic geometry of the pigments in PSII reaction centers, such secondary interactions are almost certainly present (61). These secondary interactions will be strongly modified by the formation of the P680 triplet state. The triplet state is reported to be localized on the accessory chl (38, 62). It may be possible to rationalize the triplet state bleaching observed at 684–685 nm by considering these secondary interactions.

From the above, we take that, to a first approximation, P680 is dominated by an exciton-coupled dimer analogous to the bacterial RC dimer (P850). The intense lower component of the split pair ( $Q_y^-$ ) is at 14 630  $\text{cm}^{-1}$  (683.5 nm) and has an intensity roughly twice that of a chl *a* monomer. For the expected geometry of P680 (closely antiparallel), theory requires the blue-shifted component of the exciton split  $Q_y$  band ( $Q_y^+$ ) to have low intensity. This feature should have opposite CD but similar width to the allowed band.

A plausible candidate for the  $Q_y^+$  transition in the absorbance and CD data is visible at 14 817  $\text{cm}^{-1}$  (674.9 nm), which is indicated in Figure 6a,b. The band is of similar width ( $\sim 50\text{ cm}^{-1}$ ), with an intensity in absorbance  $\sim 20\%$  of that for the 14 630  $\text{cm}^{-1}$  band and a position at the blue edge of the MCD deficit region. The exciton coupling constant inferred from the spacing is  $\sim 94\text{ cm}^{-1}$ .

The assignment of the bands at 14 630 and 14 817  $\text{cm}^{-1}$  to strongly interacting pigments within the P680 system is further supported by recent studies in our laboratories (37) of electrochromic band shifts following functional turnover ( $S_1 \rightarrow S_2$ ) in the PSII core complexes. These experiments show the pigments giving rise to the 14 630 and 14 817  $\text{cm}^{-1}$  bands to be *close* to the catalytic Mn site in the D1 protein and that the two bands respond *equivalently* to charge movements associated with S-state turnover.

Applying the transition point dipole model [e.g., (53)] to the observed exciton coupling (94  $\text{cm}^{-1}$ ) of the pair, the chl center-to-center separation in the plant special pair is 11–12 Å, slightly longer than the value of 10.0 Å observed in cyanobacteria (20).

The CD of the 14 630  $\text{cm}^{-1}$  band is the most intense in the region, with  $\Delta\epsilon_{\text{max}}/\epsilon_{\text{max}} = -4.0 \times 10^{-3}$ . The 14 817  $\text{cm}^{-1}$  band has associated a positive CD component of appropriate

width, but much smaller amplitude than the strong negative band at  $14\,630\text{ cm}^{-1}$ . Phenomenologically, such pronounced CD nonconservation (asymmetry) in the  $Q_y$  region of chl systems can be associated with dimerization, or higher oligomer formation (53, 63–65). Simple exciton coupling theory [e.g., (54)] does not predict such an effect, but chromophore interaction-induced mixing of higher energy states ( $Q_x$ ,  $B_{x,y}$ , etc) into the  $Q_y$  levels can induce CD asymmetry in the  $Q_y$  region (63, 64). However, for this effect to be substantial, the exciton coupling has to be large (hundreds of  $\text{cm}^{-1}$ ) to induce the necessary mixing. This is essentially excluded by the narrowness of the PSII core complex  $Q_y$  envelope. Further, the CD is still conservative overall, when all transitions are considered. That is, CD asymmetry in the  $Q_y$  region is balanced by correspondingly intense, complementary CD asymmetries in the  $Q_x$ , B transitions, etc.

We see no evidence of this, with weak to negligible CD at higher energies. Parallel behavior is seen for some model chl oligomer systems exhibiting comparably intense CD asymmetry in the  $Q_y$  region (65). One possible explanation is that structural effects (such as Mg ligation) arising from close chromophore association induce a large change in the nonconservative, intrinsic CD of the  $Q_y$  transitions of the interacting chromophores. There is evidence that specific chlorin–protein interactions in bacterial reaction centers produce similar effects (66).

The CD of a single chromophore may be strongly influenced by its environment in a protein, which may include a structural influence associated with neighboring chromophores. These changes may compete with or disguise the excitonic CD effects usually discussed.

**Comparison with Solubilized D1/D2/Cyt  $b_{559}$  Material.** As is evident from the comparison in Figure 6, spectral features in the reaction center region of the cores do not have a 1:1 correspondence with the spectra reported for isolated D1/D2/ $b_{559}$ . The reported CD spectra of these preparations are inverted with respect to those observed here. Our spectra have the same sign as those of earlier, more intact preparations (55, 56). This, combined with the variability of optical spectra reported with varying detergent treatments (31), suggests a significant perturbation of the chromophore interactions in D1/D2/ $b_{559}$  preparations, compared to the intact reaction center.

The reaction center region absorption envelope in functional PSII cores is narrower and red-shifted compared to the spectrum of D1/D2/ $b_{559}$  particles. We take this as evidence for substantial changes in both coupling and environment of the reaction center pigments.

D1/D2/ $b_{559}$  spectra show a weak shoulder near the position of the prominent P680 feature of the intact system. Whether this represents a minor population with 'intact' P680 chromophore geometry (see Figure 6) or some other component is as yet unclear.

**Comparison with Cyanobacterial PSII.** The absorption envelope of the intact PSII core in higher plants is substantially different from that in cyanobacteria. This is evident in 77 K spectra (67) but especially so in 4 K spectra (V. M. Masters et al., unpublished work). The difference is most pronounced near the red edge of the  $Q_y$  envelope and may reflect a significant difference of the P680 organization in plants and cyanobacteria, as discussed above. However, low-

temperature single-turnover studies on our plant PSII samples (37) show that bands involving the active branch pheophytin have very similar location and electrochromic response to those in cyanobacteria (67).

**PSII Stoichiometry.** The higher plant core complex presented here contains 4.5 Mn per 32 chlorophylls (1 Mn per 7.1 chl *a*). While a value of 4 Mn per reaction center of PSII is widely accepted, the literature data are, in fact, more equivocal. The early studies involving PSII fractions from higher plants [reviewed in (7)] suffered uncertainties concerning the fraction of total sample Mn assignable to the water splitting site. The precise chl-to-reaction center ratio in detergent-solubilized systems also suffered uncertainties.

Purified, active PSII core complexes from cyanobacteria have been available for over a decade, but the reported chl per Mn in these systems has always exceeded  $\sim 11$  and has an average of  $\sim 15$  (7, 8, 10, 14). Given that the cyanobacterial and higher plant PSII core complexes are now known to have very similar chl contents (32–34 chl *a* per reaction center), this result is puzzling, as the reported Mn content and specific catalytic activity show no obvious correlation for the cyanobacterial preparations. The total electron density volume assigned to Mn in the recently reported structure of PSII from thermophilic bacteria (20) is consistent with a Mn per reaction center stoichiometry greater than the value of 2 suggested by a chl/Mn ratio of 15.

The stoichiometry of cyt  $b_{559}$  per PSII and indeed the exact function of the cytochrome unit itself have long been under discussion. The quoted range for cyt  $b_{559}$  content in PSII is 1–2 per photosystem monomer [e.g., see (68)]. A value of 1 is confirmed for the cyanobacterial PSII core complex, by crystallography (20). This is also the value we find in the higher plant complex.

From densitometric scans of the gel patterns, we find essentially the same (within 20% uncertainty) stoichiometry of the cyt  $b_{559}$   $\alpha$ -subunit band (relative to the D1/D2, CP47, CP43 peptides) in PSII core complexes and the PSII membrane particles from which the complexes were made. This is confirmed by the MCD determination of the total cyt  $b_{559}$  signals from PSII particles. Our approach to the cyt  $b_{559}$  stoichiometry problem is novel and avoids some of the difficulties inherent in analytical procedures previously employed [see discussion in (68)]. We feel that a stoichiometry of 1.0 cyt  $b_{559}$  per reaction center is the correct figure for PSII-enriched particles derived from spinach, although the average stoichiometry for the whole thylakoid membrane is not addressed in our studies.

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## REFERENCES

1. Bricker, T. M. (1990) *Photosynth. Res.* 24, 1–13.
2. Svensson, B., Vass, I., and Styring, S. (1991) *Z. Naturforsch.* 46C, 765–776.
3. Barber, J., Nield, J., Morris, E. P., Zheleva, D., and Hankamer, B. (1997) *Physiol. Plant.* 100, 817–827.
4. Woollman, F.-A., Minai, L., and Nechushtai, R. (1999) *Biochim. Biophys. Acta* 1411, 21–85.
5. Debus, R. J. (1992) *Biochim. Biophys. Acta* 1102, 269–352.



6. Glazer, A. N. (1994) *J. Appl. Phycol.* 6, 105–112.
7. Pauly, S., and Witt, H. T. (1992) *Biochim. Biophys. Acta* 1099, 211–218.
8. Noren, G. H., Boerner, R. J., and Barry, B. A. (1991) *Biochemistry* 30, 3943–3950.
9. van Leeuwen, P. J., Nieveen, M. C., van de Meent, E. J., Dekker, J. P., and van Gorkom, H. J. (1991) *Photosynth. Res.* 28, 149–153.
10. Tang, X.-S., and Diner, B. A. (1994) *Biochemistry* 33, 4594–4603.
11. Roobol-Bóza, M., and Andersson, B. (1996) *Anal. Biochem.* 235, 127–133.
12. Hankamer, B., Nield, J., Zheleva, D., Boekema, E., Jasson, S., and Barber, J. (1997) *Eur. J. Biochem.* 243, 422–429.
13. Bricker, T. M., Morvant, J., Masri, N., Sutton, H. M., and Frankel, L. K. (1998) *Biochim. Biophys. Acta* 1409, 50–57.
14. Reifler, M. J., Chisholm, D. A., Wang, J., Diner, B. A., and Brudvig, G. W. (1998) in *Photosynthesis: Mechanisms and Effects* (Garab, G., Ed.) pp 1189–1192, Kluwer Academic Publishers, Dordrecht, The Netherlands.
15. Sugiura, M., Minagawa, J., and Inoue, Y. (1999) *Plant Cell Physiol.* 40, 311–318.
16. Sugiura, M., and Inoue, Y. (1999) *Plant Cell Physiol.* 40, 1219–1231.
17. Kuhl, H., Kruip, J., Seidler, A., Krieger-Liszkay, A., Bünker, M., Balk, D., Scheidig, A. J., and Rögner, M. (2000) *J. Biol. Chem.* 275, 20652–20659.
18. Shen, J.-R. (2000) *Biochemistry* 39, 14739–14744.
19. Zouni, A., Jordan, R., Schlodder, E., Fromme, P., and Witt, H. T. (2000) *Biochim. Biophys. Acta* 1457, 103–105.
20. Zouni, A., Witt, H.-T., Kern, J., Fromme, P., Kraub, N., Saenger, W., and Orth, P. (2001) *Nature* 409, 739–743.
21. Eshaghi, S., Andersson, B., and Barber, J. (1999) *FEBS Lett.* 446, 23–26.
22. Rhee, K.-H., Morris, E. P., Barber, J. A., and Kühlbrandt, W. (1998) *Nature* 396, 283–286.
23. Barber, J., Morris, E., and Büchel, C. (2000) *Biochim. Biophys. Acta* 1459, 239–247.
24. Michel, H., and Deisenhofer, J. (1988) *Biochemistry* 27, 1–7.
25. Svensson, B., Etchebest, C., Tuffery, P., van Kan, P., Smith, J., and Styring, S. (1996) *Biochemistry* 35, 14486–14502.
26. Schelvis, J. P. M., van Noort, P. I., Aartsma, T. J., and van Gorkom, H. J. (1994) *Biochim. Biophys. Acta* 1184, 242–250.
27. Durrant, J. R., Klug, D. R., Kwa, S. L. S., van Grondelle, R., Porter, G., and Dekker, J. P. (1995) *Proc. Natl. Acad. Sci. U.S.A.* 92, 4798–4802.
28. Greenfield, S. R., Seibert, M., and Wasielewski, M. R. (1999) *J. Phys. Chem. B* 103, 8364–8374.
29. Jankowiak, R., Rätsep, M., Picorel, R., Seibert, M., and Small, G. J. (1999) *J. Phys. Chem. B* 103, 9759–9769.
30. Nanba, O., and Satoh, K. (1987) *Proc. Natl. Acad. Sci. U.S.A.* 84, 109–112.
31. van der Vos, R., van Leeuwen, P. J., Braun, P., and Hoff, A. J. (1992) *Biochim. Biophys. Acta* 1140, 184–198.
32. Smith, P. J., Ahrling, K. A., and Pace, R. J. (1993) *J. Chem. Soc., Faraday Trans. 89*, 2863–2868.
33. Porra, R. J., Thompson, W. A., and Kriedemann, P. E. (1989) *Biochim. Biophys. Acta* 975, 384–394.
34. Gilmore, A. M., and Yamamoto, H. Y. (1991) *J. Chromatogr.* 543, 137–145.
35. Smith, P. J., and Pace, R. J. (1996) *Biochim. Biophys. Acta* 1275, 213–220.
36. Stranger, R., Dubicki, L., and Krausz, E. (1996) *Inorg. Chem.* 35, 4218–4226.
37. Masters, V. M., Smith, P. J., Krausz, E., and Pace, R. J. (2001) *J. Lumin.* (in press).
38. Kwa, S. L. S., Eijkelhoff, C., van Grondelle, R., and Dekker, J. P. (1994) *J. Phys. Chem.* 98, 7702–7711.
39. Groot, M.-L., Frese, R. N., de Weerd, F., Bromek, K., Pettersson, A., Peterman, E. J. G., van Stokkum, I. H. M., van Grondelle, R., and Dekker, J. P. (1999) *Biophys. J.* 77, 3328–3340.
40. van Dorssen, R. J., Breton, J., Plijter, J. J., Satoh, K., van Gorkom, H. J., and Ames, J. (1987) *Biochim. Biophys. Acta* 893, 267–274.
41. Alfonso, M., Montoya, G., Cases, R., Rodriguez, R., and Picorel, R. (1994) *Biochemistry* 33, 10494–10500.
42. Chang, H. C., Jankowiak, R., Yocum, C. F., Picorel, R., Alfonso, M., Seibert, M., and Small, G. J. (1994) *J. Phys. Chem.* 98, 7717–7724.
43. Piepho, S. B., and Schatz, P. N. (1983) *Group Theory in Spectroscopy, with Applications to Magnetic Circular Dichroism*, Wiley, New York.
44. Kobayashi, M., Wang, Z.-Y., Yoza, K., Umetsu, M., Konami, H., Mimuro, M., and Nozawa, T. (1996) *Spectrochim. Acta, Part A* 51, 585–598.
45. Cheesman, M. R., Greenwood, C., and Thomson, A. J. (1991) *Adv. Inorg. Chem.* 36, 201–255.
46. Sievers, G., Gadsby, P. M. A., Peterson, J., and Thomson, A. J. (1983) *Biochim. Biophys. Acta* 742, 637–647.
47. Foote, N., Peterson, J., Gadsby, P. M. A., Greenwood, C., and Thomson, A. J. (1984) *Biochem. J.* 223, 369–378.
48. Moore, G. R., Williams, R. J. P., Peterson, J., Thomson, A. J., and Mathews, F. S. (1985) *Biochim. Biophys. Acta* 829, 83–96.
49. Ahrling, K. A., and Styring, S. (2000) in *Probing Photosynthesis—Mechanisms, Regulation and Adaptation* (Yunus, M., Pathre, U., and Mohanty, P., Eds.) pp 148–163, Taylor and Francis, London.
50. Miller, A.-F., and Brudvig, G. W. (1991) *Biochim. Biophys. Acta* 1056, 1–18.
51. Boussac, A., Girerd, J.-J., and Rutherford, A. W. (1996) *Biochemistry* 35, 6984–6989.
52. Boussac, A., Un, S., Horner, O., and Rutherford, A. W. (1998) *Biochemistry* 37, 4001–4007.
53. Houssier, C., and Sauer, K. (1970) *J. Am. Chem. Soc.* 92, 779–791.
54. Koolhaas, M. H. C., van der Zwan, G., Frese, R. N., and van Grondelle, R. (1997) *J. Phys. Chem. B* 101, 7262–7270.
55. van Dorssen, R. J., Plijter, J. P., Dekker, J. P., den Ouden, A., Ames, J., and van Gorkom, H. J. (1987) *Biochim. Biophys. Acta* 890, 134–143.
56. Canaan, O. D., and Sauer, K. (1978) *Biochim. Biophys. Acta* 501, 545–551.
57. Carbonera, D., Giacometti, G., and Agostini, G. (1994) *FEBS Lett.* 343, 200–204.
58. Hillmann, B., Brettel, K., van Mieghem, F., Kamlowski, A., Rutherford, A. W., and Schlodder, E. (1995) *Biochemistry* 34, 4814–4827.
59. van Mieghem, F., Brettel, K., Hillmann, B., Kamlowski, A., Rutherford, A. W., and Schlodder, E. (1995) *Biochemistry* 34, 4798–4813.
60. Nozawa, T., Kobayashi, M., Wang, Z.-Y., Itoh, S., Iwaki, M., Mimuro, M., and Satoh, K. (1995) *Spectrochim. Acta, Part A* 51, 125–134.
61. Dekker, J. P., and van Grondelle, R. (2000) *Photosynth. Res.* 63, 195–208.
62. van Mieghem, F. J. E., Satoh, K., and Rutherford, A. W. (1991) *Biochim. Biophys. Acta* 1058, 379–385.
63. Scherz, A., and Parson, W. W. (1984) *Biochim. Biophys. Acta* 766, 666–678.
64. Koolhaas, M. H. C., van der Zwan, G., van Mourik, F., and van Grondelle, R. (1997) *Biophys. J.* 72, 1828–1841.
65. Tamiaki, H., Kubo, M., and Oba, T. (2000) *Tetrahedron* 56, 6245–6257.
66. Mar, T., and Gingras, G. (1995) *Biochemistry* 34, 9071–9078.
67. Stewart, D. H., Nixon, P. J., Diner, B. A., and Brudvig, G. W. (2000) *Biochemistry* 39, 14583–14594.
68. Stewart, D. H., and Brudvig, G. W. (1998) *Biochim. Biophys. Acta* 1367, 63–87.
69. Zheleva, D., Sharma, J., Panico, M., Morris, H. R., and Barber, J. (1998) *J. Biol. Chem.* 273, 16122–16127.